



## Salt base solution

$(\text{NH}_4)_2\text{SO}_4$	1.30	g
$\text{KH}_2\text{PO}_4$	0.28	g
$\text{MgSO}_4 \times 7 \text{ H}_2\text{O}$	0.25	g
$\text{CaCl}_2 \times 2 \text{ H}_2\text{O}$	0.07	g
$\text{MnCl}_2 \times 4 \text{ H}_2\text{O}$	1.80	mg
$\text{Na}_2\text{B}_4\text{O}_7 \times 10 \text{ H}_2\text{O}$	4.50	mg
$\text{ZnSO}_4 \times 7 \text{ H}_2\text{O}$	0.22	mg
$\text{CuCl}_2 \times 2 \text{ H}_2\text{O}$	0.05	mg
$\text{Na}_2\text{MoO}_4 \times 2 \text{ H}_2\text{O}$	0.03	mg
$\text{VO}_2 \times n \text{ H}_2\text{O}$	0.03	mg
$\text{CoSO}_4 \times 7 \text{ H}_2\text{O}$	0.01	mg
Distilled water	1000.00	ml

Adjust pH to 2.5 with 10 N  $\text{H}_2\text{SO}_4$ .