



## Salt solution

$(\text{NH}_4)_2\text{SO}_4$	1.00	g
KCl	0.50	g
$\text{MgSO}_4 \times 7 \text{H}_2\text{O}$	0.50	g
$\text{CaCl}_2 \times 2 \text{H}_2\text{O}$	0.20	g
$\text{CaCO}_3$	0.20	g
$\text{FeCl}_3 \times 6 \text{H}_2\text{O}$	5.00	mg
Demineralized water	1000.00	ml